



ENROLL US!

We Want to Be a Partner in EPA's
National Partnership for Environmental Priorities

IDENTIFYING INFORMATION

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Date: December 7, 2005

PARTNER AGREEMENT

Our organization is choosing to become a partner in EPA's National Partnership for Environmental Priorities. Our goal is to reduce the quantity of one or more Priority Chemicals currently found in our products, processes, or releases using techniques such as source reduction, recycling, or other materials management practices. In this enrollment application, we identify one or more voluntary goals that we believe we can achieve as partners in this program. The voluntary goal(s) provided below is an initial estimate and may change over time. We may revise our goal(s) or withdraw from the program at any time. If/when we choose to revise our goals or withdraw from the program, we will notify EPA.

GOAL #1. Chemical Name: Lead **CASRN:** 7439-92-1

Narrative description of proposed project: Our corporate goal is to have 95% of all new equipment lead solder-free by the end of 2008. This goal will reduce 1050 pounds of lead from printed circuit boards used in Kodak equipment by the end of 2008. We plan on accomplishing this goal by (1) setting a corporate standard for lead to ensure that new equipment is lead solder-free, (2) implementing lead-free solder design requirements, and (3) pushing redesign and new technology conversion with equipment designers.

How we will measure success: We will measure success by comparing the amount of lead used before and after the project.

1a. Our voluntary **source reduction** goal for Chemical #1 is to reduce the amount of this chemical generated/used from a baseline amount of 1850 pounds in December, 2004 (month/year) to a reduced amount of 800 pounds generated/used by December, 2008 (month/year).

1b. To accomplish this goal, we will use the following source reduction options (check all that apply):

<input checked="" type="checkbox"/> Equipment or technology modifications.	<input checked="" type="checkbox"/> Process or procedure modifications.
<input checked="" type="checkbox"/> Reformulation or redesign of products.	<input checked="" type="checkbox"/> Substitution of less toxic raw materials.
<input checked="" type="checkbox"/> Improvements in inventory control.	<input type="checkbox"/> Improvements in maintenance/housekeeping practices.
<input type="checkbox"/> Other (describe): _____	

2a. In addition to, or in lieu of using source reduction methods, our voluntary **recycling or recovery** goal for Chemical # 1 is to increase the recycled or recovered quantity of this chemical from a baseline amount of _____ pounds in _____ (month/year) to an increased quantity of _____ pounds by _____ (month/year).

2b. To accomplish this recycling or recovery goal, we will use the following options (check all that apply):

<input type="checkbox"/> Direct use/reuse in a process to make a product.
<input type="checkbox"/> Processing the waste to recover or regenerate a usable product.
<input type="checkbox"/> Using/reusing waste as a substitute for a commercial product.
<input type="checkbox"/> Other (describe): _____

SUPPLEMENTAL GOAL SHEET: NATIONAL PARTNERSHIP FOR ENVIRONMENTAL PRIORITIES

GOAL # 2 . Chemical Name: Methylene Chloride **CASRN:** 75-09-2

Narrative description of proposed project: As part of our Responsible Growth Program, Kodak established a set of worldwide Health, Safety, and Environment Goals designed to reduce emissions, conserve resources, and enhance the company's product stewardship and employee safety. As part of this program, Kodak established a worldwide goal to reduce methylene chloride emissions by 35%.

How we will measure success: We will measure success by comparing the amount of methylene chloride used before and after the project.

1a. Our voluntary **source reduction** goal for Chemical # is to reduce the amount of this chemical generated/used from a baseline amount of 937,000 pounds in December, 2002 (month/year) to a reduced amount of 600,000 pounds generated/used by December, 2008 (month/year).

1b. To accomplish this goal, we will use the following source reduction options (check all that apply):

<u> X </u>	Equipment or technology modifications.	<u> X </u>	Process or procedure modifications.
<u> </u>	Reformulation or redesign of products.	<u> </u>	Substitution of less toxic raw materials.
<u> </u>	Improvements in inventory control.	<u> </u>	Improvements in maintenance/housekeeping practices.
<u> </u>	Other (describe):		

2a. In addition to, or in lieu of using source reduction methods, our voluntary **recycling or recovery** goal for Chemical # ____ is to increase the recycled or recovered quantity of this chemical from a baseline amount of _____ pounds in _____ (month/year) to an increased quantity of _____ pounds by _____ (month/year).

2b. To accomplish this recycling or recovery goal, we will use the following options (check all that apply):

_____ Direct use/reuse in a process to make a product.
 _____ Processing the waste to recover or regenerate a usable product.
 _____ Using/reusing waste as a substitute for a commercial product.
 _____ Other (describe): _____

GOAL # . **Chemical Name:**_____ **CASRN:**_____

Narrative description of proposed project: _____

How we will measure success:

1a. Our voluntary **source reduction** goal for Chemical # ____ is to reduce the amount of this chemical generated/used from a baseline amount of _____ pounds in _____ (month/year) to a reduced amount of _____ pounds generated/used by _____ (month/year).

1b. To accomplish this goal, we will use the following source reduction options (check all that apply):

_____ Equipment or technology modifications.	_____ Process or procedure modifications.
_____ Reformulation or redesign of products.	_____ Substitution of less toxic raw materials.
_____ Improvements in inventory control.	_____ Improvements in maintenance/housekeeping practices.
_____ Other (describe):	

2a. In addition to, or in lieu of using source reduction methods, our voluntary **recycling or recovery** goal for Chemical # _____ is to increase the recycled or recovered quantity of this chemical from a baseline amount of _____ pounds in _____ (month/year) to an increased quantity of _____ pounds by _____ (month/year).

2b. To accomplish this recycling or recovery goal, we will use the following options (check all that apply):

☐ Direct use/reuse in a process to make a product.
☐ Processing the waste to recover or regenerate a usable product.
☐ Using/reusing waste as a substitute for a commercial product.
☐ Other (describe): _____

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